

2.2 Review

2.2 a. What are the main nutrients found in food? 2.2 b. How can carbohydrates, lipids, and proteins be detected in food?

Sugar, starches, lipids, and proteins are found in food. These nutrients supply the energy and resources needed by our cells to function correctly.

Tests	on for	od rea	uire sr	pecific	chemical	indicators.
1 0010	01110	Ja i Oq		3001110	onionnioai	maioatoro.

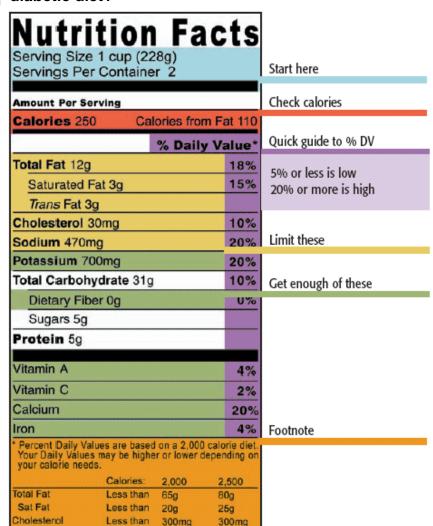
- *Starch is identified with -
- *Sugar is identified with -
- *Protein is identified with -

Sodium

Total Carbohydrate

Dietary Fiber

2.2 c. What types of foods supply sugar, starch, proteins, and lipids? 2.2 d. How can food labels be used to evaluate dietary choices? 2.2 f. What are basic recommendations for a diabetic diet?



2,400mg

375g

Less than 2,400mg

300g

25g

- *Sugars and starches come from carbohydrates. Grains, fruits, veggies, dairy, sweets.
- *Proteins come from dairy, meat, beans, nuts, eggs
- *Lipids come from saturated, unsaturated, and trans fats. Meat, oils, nuts, veggies, dairy.
- *Food labels give us an idea of how much to eat based on a 2000 calorie diet. Serving sizes and daily values are guides.
- *Diabetic diets are based on high protein low carbohydrate/sugar foods.
- *Would this food label represent a food that would be recommended or avoided? Explain your choice.

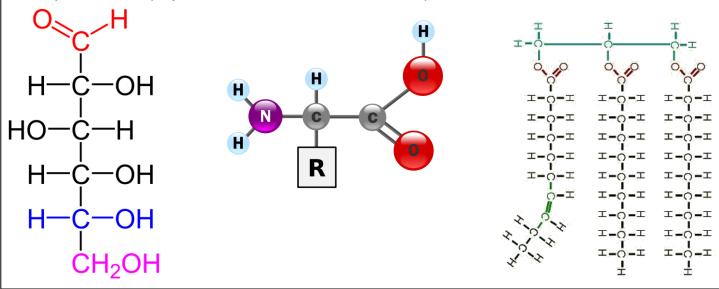
2.2 e. What role do basic nutrients play in the function of the human body?

Carbohydrates and sugar are used as the body's primary energy source. Glucose is used to produce molecules of ATP. Proteins are used to build muscle, form enzymes and hormones, and repair cellular parts. Lipids are also an energy source. They provide more energy than carbohydrates but are not used until the carbohydrate supply is gone. They provide the material for the cell membrane and fat-soluble vitamins.

2.2 g. What are the main structural components of carbohydrates, proteins, and lipids?

MONOMER	POLYMER
Monosaccharide (glucose, fructose, galactose)	Carbohydrate (C, H, O)
Amino Acid (20 different)	Protein (amine, variable group, carboxyl group)
Glycerol + Fatty Acid Chain	Saturated/Unsaturated fat (Triglyceride)

Identify each of the polymers below and the elements/components in the table.



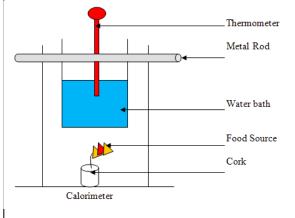
2.2 h. What is dehydration synthesis and hydrolysis? 2.1 i. How do dehydration synthesis and hydrolysis relate to harnessing energy from food?

Dehydration Synthesis "Remove water and come together"	Comparison	Hydrolysis "Split with water"
The reaction that builds macromolecules. As 2 molecules join to form 1, they give off a water molecule.	They are exact opposites.	The reaction that splits macromolecules, typically during digestion. Each water molecule can break one bond.
This reaction <i>requires</i> energy.		This reaction <i>releases</i> energy.

Label the following reactions as dehydration synthesis and hydrolysis.

2.2 j. How is the amount of energy in food determined?

The energy in food is determined by calorimetry ("energy measurement"). Food is burned to break the chemical bonds. The heat released by the bonds breaking is captured in water. The change in mass and the change in temperature are used to determine the cal/g in the food sample.



Calories in Food Sample =
$$(600g)(Change in Water Temperature (°C)) \left(\frac{1 Calorie}{Kg^{\circ}C}\right) \left(\frac{1 Kg}{1000g}\right)$$

Calories per Gram (Cal/g) =
Calories in Food Sample (Cal) + Change in the Mass of the sample (g)